# GLYCOZOLIDAL, A NEW CARBAZOLE ALKALOID FROM GLYCOSMIS PENTAPHYLLA

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In continuation of our work on carbazole alkaloids of the Rutaceae (1), we now report the structure of another new carbazole alkaloid, glycozolidal (1), from the root of *Glycosmis pentaphylla* (Retz) DC., a reputed Indian medicinal plant known to furnish carbazole alkaloids (2,3).

Glycozolidal (1) was found to be homogeneous by tlc and ms. It gave a picrate, formed a 2,4 dinitrophenyl hydrazone, and reduced ammonical AgNO<sub>3</sub> solution, indicating the presence of an aldehyde function. Analytical data indicated the presence of two methoxy groups. The uv spectrum of the compound was characteristic of 3-formyl carbazole (4). The ir spectrum showed it to be an aromatic compound with -NH. aldehyde and aromatic ether functions. In the <sup>1</sup>H-nmr spectrum, the C-4 and C-5 protons of the carbazole nucleus, which appeared at a lower field than the other aromatic protons, are readily recognized (5). Thus, the signals at  $\delta$  8.4 and  $\delta$  7.6 were assigned to C-4 and C-5 hydrogens, respectively. The proton at C-4 was deshielded due to the close proximity of the aldehyde function at C-3. Moreover, the C-4 proton was not ortho- or meta-coupled, suggesting substitutions at positions 3 and 2. The C-5 proton was also not ortho-coupled which suggested substitution at position 6.

From these observations, the structure of glycozolidal was formulated as 1. This has been confirmed by oxidation of glycozolidine (2) with 2,3-dichloro-5,6-dicyanobenzoquinone when a compound identical in all respects with natural glycozolidal was obtained.



1  $R_1 = R_3 = OCH_3$ ;  $R_2 = CHO$ 2  $R_1 = R_3 = OCH_3$ ;  $R_2 = CH_3$ 

Evidently glycozolidal could be considered as the biological oxidation product of glycozolidine where the Ar-Me group has been oxidized to Ar-CHO group. This type of oxidation has its precedence in the genus Murraya (6), which is taxonomically close to G. pentaphylla.

## **EXPERIMENTAL**

PLANT MATERIAL.—G. pentaphylla was collected locally and identified by Dr. S.R. Das, Survey Officer, Regional Research Institute (AY), Calcutta, and a voucher specimen has been deposited at Bose Institute.

ISOLATION OF GLYCOZOLIDAL (1).—Airdried, finely powdered roots (2 kg) of *G. pentaphylla* were extracted with petroleum ether (40-60°) in a Soxhlet for 48 h. After solvent evaporation, the residue was taken in a little  $C_6H_6$  and chromatographed over silica gel (450 g). The column was eluted with petroleum ether, petroleum ether- $C_6H_6$  (1:1),  $C_6H_6$ , CHCl<sub>3</sub>, and MeOH, in succession. The petroleum ether- $C_6H_6$  (1:1) eluate furnished a solid, which, on crystallization from  $C_6H_6$ , furnished glycozolidal (150 mg).

GLYCOZOLIDAL (1).—mp 185°,  $C_{15}H_{13}NO_3$ (M<sup>+</sup>, 255); uv  $\lambda$  max (ErOH) 235, 248, 303, 340 nm (log  $\epsilon$  4.4, 4.2, 4.2, 4.06); ir  $\nu$  max (KBr) 3500 (-NH), 1675 (aromatic aldehyde) 1600, 1595 (aromatic residue) 1220, 1200 cm<sup>-1</sup> (aromatic ether); <sup>1</sup>H nmr (60 MHz, CDCl<sub>3</sub>)  $\delta$  9.9 (s, 1H, CHO) 8.8 (brs, 1H, NH, confirmed by D<sub>2</sub>O exchange), 8.4 (s, 1H, H-4), 7.6 (d, 1H, J=2 Hz, H-5) 6.85 (dd, 1H, J=7.0 Hz and 2.0 Hz, H-7) 7.3 (d, 1H, J=7.0 Hz, H-8) in addition to two singlets at  $\delta$  3.9 (3H) and 3.7 (3H) for two methoxy groups. OXIDATION OF GLYCOZOLIDINE (2).— Glycozolidine (100 mg) was dissolved in dry  $C_6H_6$  and stirred with 2,3-dichloro-5-6-dicyanobenzoquinone at room temperature for 30 min. The reaction mixture was washed successively with 5% HCl and H<sub>2</sub>O and then dried. The residue was taken in a little  $C_6H_6$  and chromatographed on silica gel (10 g). Elution with  $C_6H_6$  furnished a compound (25 mg) found to be identical with glycozolidal by direct comparison in all respects (mp, mmp, tlc, uv, ir).

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